Perkin Communications

Intramolecular (3 + 2) Cycloadditions

Michael P. Collins, John Mann, John Mann, John Mann, John Mann, John Mann, Migel Capps Mand Harry Finch

^a Department of Chemistry, Reading University, Whiteknights, Reading RG6 2AD, UK

^b Glaxo Group Research Ltd., Ware, Herts SG12 0DJ, UK

1-(1-Hydroxy-2-trimethylsilylmethylprop-2-enyl)-2-(4-methoxybut-3-enyl)-4-methoxybenzene, prepared by a short synthesis, underwent (3 + 2) cycloaddition to form predominantly the *cis*-fused tricycle 3,7-dimethoxy-1-methylene-2,3,3a,4,5,9b-hexhydro-1*H*-benz[*e*]indene upon treatment with TiCl₄ or triflic anhydride.

There are numerous reported intermolecular (3 + 2) cyclo-additions involving allyl cations, but only a small number of examples that involve the corresponding intramolecular process. We have investigated the transformation shown in Scheme 1, and can report that it is both reasonably efficient and stereoselective.

MeO TICL4 N-methylaniline
$$-20^{\circ}\text{C}$$
 MeO TICL4 N-methylaniline -20°C H MeO TICL4 N-methylaniline -20°C H MeO TICL4 N-methylaniline -20°C H MeO TICL4 N-methylaniline -20°C Ratio 2:3 -20°C Ratio 2:3 -20°C Ratio 2:3 -20°C SMe 1:3 -20°C Neo -20°C

Scheme 1

The various substrates 1 for the cycloadditions were prepared by the route shown in Scheme 2, and these were then treated with $TiCl_4$ in the presence of N-methylaniline in dichloromethane at $-20\,^{\circ}$ C, or with triflic anhydride in dichloromethane at $-78\,^{\circ}$ C.† Careful examination of the 1 H NMR spectra (coupling and NOE data), revealed that in each instance the major product was the cis-fused cycloadduct with a synalkoxy or thioalkoxy group 2. Small amounts of the alternative anti-product 3 were also obtained. Extensive investigation of the cycloaddition with 1a (up to a 1 g scale), showed that the ratio of 2a:3a was apparently independent of the stereochemical

Scheme 2 Reagents and conditions: i, LiAlH₄ (65–75%); ii, bromine (70%); iii, pyridinium chlorochromate (80–90%); iv, Ph₃P=CHX (30–75%) (X = OMe, SMe, OCH₂Ph); v, BuLi then DMF (75%); vi, 3-trimethylsilyl-2-bromomagnesiopropene (70–95%)

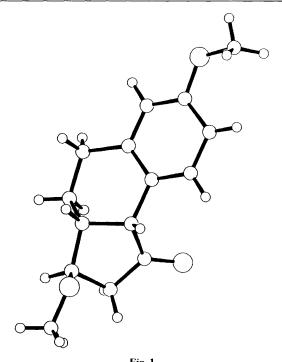


Fig. 1

composition of the enol ethers 1a (trans: cis was varied from 1:0 to 1:1). This is probably a reflection of the relative stabilities of the final products arising from a common intermediate 4. Confirmation of the all-syn stereochemistry of 2a was obtained through an X-ray crystallographic study on the ketone 5 (vide infra). The ORTEP representation is shown in Fig. 1.³

The tricycles 2 are potentially useful intermediates for the construction of estrogens and benzophenanthridine alkaloids, and some preliminary transformations of 2a have been investigated. Ozonolysis yielded the ketone 5 (55%), which was converted into the oxime 6 (70%). Attempted Beckmann rearrangement using a range of reagents resulted in fragmentation and obtention of the nitrile 7 (75%). Finally, reaction of 5 with BBr₃ in dichloromethane at -10 °C, followed by

[†] Typical experimental procedure. N-Methylaniline (2 equiv.) was added to titanium tetrachloride (1.8 equiv. of a 1M solution in dichloromethane) in dry dichloromethane at 0 °C. After 0.5 h the solution was cooled to -20 °C, and the enol ether (or thioether) 1 (1 equiv.) in dichloromethane was added, and the reaction mixture stirred at the same temperature for 1 h. Ether was added, followed by 1M HCl, the organic layer was collected and the products purified by flash chromatography.

treatment with an ethanolic solution of silver nitrate,⁵ yielded the enone 8. This should allow access to the *trans*-fused tricycle. Further experiments to establish the synthetic utility of the tricycles 2 are underway.

Acknowledgements

We thank Glaxo and the SERC for a CASE award to M. P. C.,

and Dr. Oliver Howarth (University of Warwick NMR service) for numerous NMR investigations.

References

- 1 (a) R. Noyori and M. Yamakawa, *Tetrahedron Lett.*, 1978, 4823; (b) R. L. Danheiser, D. J. Carini and A. Basak, *J. Am. Chem. Soc.*, 1984, **103**, 1604; (c) D. L. Boger and C. E. Brotherton, *J. Am. Chem. Soc.*, 1984, **106**, 805; (d) B. M. Trost and D. M. T. Chan, *J. Am. Chem. Soc.*, 1983, **105**, 2315.
- 2 (a) J. Ipaktschi and G. Lauterbach, Angew. Chem., 1986, 25, 354; (b)
 R. J. Giguere, S. M. Duncan, J. M. Bean and L. Purvis, Tetrahedron Lett., 1988, 6071; (c)
 R. J. Giguere, S. M. Tassely and M. I. Rose, Tetrahedron Lett., 1990, 4577; (d)
 H. M. R. Hoffmann, U. Eggert, U. Gibbels, K. Kiesel, O. Koch, R. Lies and J. Rabe, Tetrahedron, 1988, 44, 3899.
- 3 W. thank Dr. M. G. B. Drew for these X-ray studies which will be reported in detail in our full paper.
- 4 These reagents included: i, PCl₅, CHCl₃ (-20 °C); ii, SOCl₂, CCl₄ (0 °C); iii, trimethylsilyl polyphosphate, CH₂Cl₂ (RT); iv, tosyl chloride, pyridine (-20 °C); and v, hν (254 nm, 12 W low-pressure lamp), MeOH.
- 5 J. Kueh, M. Meller, and G. Pattenden, J. Chem. Soc., Perkin Trans. 1, 1981, 1052.

Paper 0/04621C Received 15th October 1990 Accepted 31st October 1990